

DRAFT OECD GUIDELINE FOR THE TESTING OF CHEMICALS

The Cytosensor Microphysiometer Test Method: An *in vitro* Method for Identifying Ocular Corrosive and Severe Irritant Chemicals as well as Chemicals not Classified as Ocular Irritants

INTRODUCTION

1. The Cytosensor Microphysiometer (CM) test method is an *in vitro* test method that can be used under certain circumstances and with specific limitations for hazard classification and labelling of chemicals for eye corrosion/irritation, according to the United Nations (UN) Globally Harmonized System of Classification and Labelling of Chemicals (GHS) (1). For the purpose of this Test Guideline, severe irritants are defined as chemicals that cause tissue damage in the eye following test chemical administration that is not reversible within 21 days or causes serious physical decay of vision, while ocular corrosives are chemicals that cause irreversible tissue damage to the eye. These chemicals are classified as UN GHS Category 1. Chemicals not classified as an eye irritant are defined as those that do not meet the requirements for classification as UN GHS Category 1, 2A or 2B. While the CM test method is not considered valid as a complete replacement for the *in vivo* rabbit eye test, the CM is recommended for use as part of a tiered testing strategy for regulatory classification and labelling. Thus, the CM is recommended as an initial step within a Top-Down approach to identify ocular corrosives/severe irritants (UN GHS Category 1), as well as an initial step within a Bottom-Up approach to identify chemicals that do not require classification for eye corrosion/irritation (UN GHS Not Classified), specifically for limited types of chemicals (*i.e.* water soluble surfactants and surfactant-containing mixtures) (4)(5).

2. It is currently generally accepted that, in the foreseeable future, no single *in vitro* eye irritation test will be able to replace the *in vivo* Draize eye test to predict across the full range of irritation for different chemical classes. However, strategic combinations of several alternative test methods within a (tiered) testing strategy may be able to replace the Draize eye test (5). The Top-Down approach (5) is designed to be used when, based on existing information, a chemical is expected to have high irritancy potential, while the Bottom-Up approach (5) is designed to be used when, based on existing information, a chemical is expected not to cause sufficient eye irritation to require a classification. Based on the prediction model detailed in paragraph 29, the CM test method can identify substances and mixtures within a limited applicability domain as ocular corrosives/severe irritants (UN GHS Category 1) or as not classified for eye corrosion/irritation (UN GHS Not Classified) without any further testing. Therefore, the CM test method may be used as a partial replacement test for *in vivo* acute eye irritation/corrosion testing to determine the eye irritancy/corrosivity of chemicals, following the sequential testing strategy of TG 405 (6). However, a chemical that is not predicted as ocular corrosive/severe irritant or as not classified for eye corrosion/irritation with the CM test method would require additional testing (*in vitro* and/or *in vivo*) to establish a definitive classification. The CM is so far the only validated *in vitro* test method that can be used to identify chemicals not classified as eye irritants but it is not considered adequately valid for the identification of mild or moderate ocular irritants (*i.e.* UN GHS Categories 2A and 2B).

3. The purpose of this Test Guideline is to describe the procedures used to evaluate the potential for ocular corrosivity or irritancy of a test chemical as measured by its ability to induce changes in cellular metabolism which occur after chemical exposure. The CM test method estimates the decrease in metabolic rate of the cells exposed to the test chemical by measuring the rate of change in pH of the medium (acidity) per unit time as compared to the basal metabolic state. The reduction of the metabolic rate of the exposed cells can be used to estimate the ocular toxicity potential of a test chemical. Annex I provides diagrams of the operating components of the CM. The capsule insert

1 described in the Cytosensor Manual and which can be seen in the second diagram of Annex I must
2 not be used in the assay (8).

3
4 4. Performance Standards (provided in Annex III) have been developed to facilitate the
5 validation of new similar test methods and allow for timely amendment of this test guideline so that
6 new similar test methods can be added if the Performance Standards are met.

7
8 5. Definitions are provided in Annex II.

9 10 **INITIAL CONSIDERATIONS AND LIMITATIONS**

11
12 6. This Test Guideline is based on INVITTOX protocol No. 130 (7) that has been evaluated in
13 an international validation study by the European Centre for the Validation of Alternative Methods
14 (ECVAM) (8), in collaboration with the US Interagency Coordinating Committee on the Validation
15 of Alternative Methods (ICCVAM) (9) and the Japanese Center for the Validation of Alternative
16 Methods (JaCVAM).

17
18 7. The CM test method is not recommended for the identification of mild/moderate irritant
19 chemicals (substances and mixtures), *i.e.* GHS Cat. 2A/2B, as demonstrated by the validation study
20 (4, 8).

21
22 8. The CM test method is used for the testing of only water soluble chemicals (substances and
23 mixtures) as well as non water-soluble solids, viscous chemicals or suspensions that maintain
24 uniformity during analysis time, *i.e.* that do not settle or separate into more than one phase. However,
25 different solubility criteria apply depending on the use given to the test method and the classification
26 system used. More specifically, the following solubility criteria apply:

- 27 i) in order to identify a chemical as "not classified for eye irritation" (UN GHS Not
28 Classified) the test chemical must form a single phase solution/suspension in Low-
29 Buffered Medium during analysis time at a concentration > 10 mg/mL for the UN GHS
30 classification and labelling systems (C&L) (1 dose above the cut-off of 10 mg/mL), or at
31 a concentration > 80 mg/mL for the U.S. EPA C&L (1 dose above the cut-off of 80
32 mg/mL). If at the maximum achievable concentration ≥ 10 mg/mL (UN GHS C&L) or \geq
33 80 mg/mL (U.S. EPA C&L) the test chemical still induces a relative cellular metabolic
34 rate > 50%, it can still be identified as UN GHS Not Classified (MRD₅₀ must be reported
35 as > 10 mg/mL) or as U.S. EPA Category IV (MRD₅₀ must be reported as > 80 mg/mL),
36 respectively.
- 37 ii) in order to identify a chemical as "inducing serious eye damage" (UN GHS Category 1)
38 or as an "ocular corrosive or severe irritant" (U.S. EPA Category I) the test chemical
39 must form a single phase solution/suspension in Low-Buffered Medium during analysis
40 time at a concentration ≥ 2 mg/mL (1 dose above the cut-off of 2 mg/mL). However, if
41 the test chemical forms a single phase solution/suspension in Low-Buffered Medium
42 during analysis time only at a concentration ≤ 2 mg/mL, but it already induces a relative
43 metabolic rate < 50% at that concentration (MRD₅₀ < 2 mg/mL), it can still be classified
44 as UN GHS Category 1 or as U.S. EPA Category I.
- 45 iii) if a test chemical only forms a single phase solution/suspension in Low-Buffered
46 Medium during analysis time at a concentration < 10 mg/mL (UN GHS) or < 80 mg/mL
47 (U.S. EPA C&L) and the relative cellular metabolic rate is still > 50% at that

1 concentration, the test chemical should be declared as "unsuitable for testing" for the
2 specific classification(s) system(s).
3

4 9. Other identified limitations of the test method are based on false negative and false positive
5 rates. When used as an initial step within a Top-Down approach to identify ocular corrosives/severe
6 irritants (UN GHS Category 1), the CM test method is considered suitable for water soluble
7 chemicals (substances and mixtures). When testing water soluble chemicals the false positive rate for
8 the CM test method ranged from 2% (1/48; UN GHS) to 8.5% (4/47; U.S. EPA) and the false
9 negative rate ranged from 20.5% (7/34; UN GHS) to 26.66% (8/30; U.S. EPA) when compared to *in*
10 *vivo* results.
11

12 10. When used as an initial step within a Bottom-Up approach to identify chemicals that do not
13 require classification for eye corrosion/irritation (UN GHS Not Classified; U.S. EPA Category IV),
14 the CM test method is considered suitable only for water soluble surfactants and surfactant containing
15 mixtures. When testing water soluble surfactants and surfactant containing mixtures, the false
16 negative rate for the CM test method ranged from 0% (0/28; UN GHS) to 2% (1/46; U.S. EPA) and
17 the false positive rate ranged from 50% (3/6; U.S. EPA) to 68% (17/25; UN GHS) when compared to
18 *in vivo* results. Because of high false negative and false positive rates for the CM test method when
19 testing water soluble non-surfactant chemicals, it is not recommended for identifying these types of
20 chemicals as not requiring ocular hazard classification. Surfactant-containing pesticide mixtures were
21 not included in the test method validation chemical set and therefore negative results (UN GHS Not
22 Classified, U.S. EPA Category IV) obtained with this type of mixtures should be considered with
23 caution and substantiated by further supporting information.
24

25 11. The current applicability domain might be increased in some cases, but only after analyzing
26 an expanded data set of studied test chemicals, preferably acquired through testing (4). This Test
27 Guideline will be updated accordingly as new information and data are considered.
28

29 **PRINCIPLE OF THE TEST**

30

31 12. The CM test method is a cytotoxicity and cell-function based *in vitro* assay that is performed
32 on a sub-confluent monolayer of adherent mouse L929 fibroblasts cultured in a sensor chamber using
33 a pH-meter to detect changes in acidity. The L929 cells were selected because they readily attach to
34 the polycarbonate filter membrane of the transwell and are easy to grow in continuous culture.
35 Although data from the use of other cell types, e.g. normal human keratinocytes, has been reported,
36 virtually all safety studies conducted with the CM instrument since the early 1990's have used L929
37 cells. Use of normal human keratinocytes with the transwell is not recommended. Mechanistically,
38 the CM test method is intended to model the cytotoxic action of an irritant chemical on the cell
39 membranes of the corneal and conjunctival epithelium where the test chemical would reside in an *in*
40 *vivo* exposure (10).
41

42 13. The CM estimates the metabolic rate of a population of cells maintained in low volume flow-
43 through chambers by measuring the rate of excretion of acid by-products and the resulting decrease
44 in pH of the surrounding medium. The metabolic rate is determined indirectly by the number of
45 protons excreted into Low-Buffered Medium (for composition see paragraph 17) (change in pH) per
46 unit time. The pH-meter forms the bottom of the flow through chamber and serves as a very sensitive
47 and stable pH meter.
48

49 14. During the course of an experiment, test samples, prepared as dilution series of a test
50 chemical, are introduced in order of increasing concentration to flow-through chambers containing
51 the cells. Therefore, in the CM test method the same cell population is exposed progressively to

1 increasing concentrations of the test chemical. The cells cultured in the chamber are exposed to the
2 test chemical for a short period of time, followed by a rinse step with Low-Buffered Medium (for
3 composition see paragraph 18) to remove the test chemical. Finally the flow is stopped and the
4 change in pH is measured. All rate of acidification measurements are made on washed cells. These
5 three steps are repeated with increasing concentrations of the test chemical until either the highest
6 testable concentration has been used or until the population of cells is severely damaged and the
7 metabolic rate has declined to effectively zero.

8
9 15. The rate of change in pH per unit time becomes the metabolic rate of the population of cells.
10 If a test chemical causes cytotoxicity to this population of cells it is assumed that the metabolic rate
11 will fall. A transient up-regulation of glucose metabolism can occur if the cells need energy to
12 maintain their integrity in the face of a mild biochemical insult, but it soon falls below the basal level
13 if exposure to the cytotoxic chemical is prolonged or intensified (higher concentration). The
14 concentration of test chemical that leads to a 50% decline in the basal metabolic rate of the
15 population (MRD₅₀; metabolic rate decrement of 50%) is the parameter used to measure the cytotoxic
16 effect of the test chemical on the test system (L929 mouse fibroblast cells). The MRD₅₀ value
17 (mg/mL) for each test chemical is calculated from a concentration response curve (see paragraph 28),
18 and is used to provide a measure of the ocular irritancy potential of the test chemical.

19
20 16. Recovery is an important part of a test chemical's toxicity profile that is also assessed by the
21 *in vivo* ocular irritation test. The CM test method is non-invasive, thus it could also be used for the
22 determination of recovery of the cells from toxic insult. Additional data, preferably acquired by
23 further testing, would be required to confirm this usefulness (8). This Test Guideline will be updated
24 accordingly as new information and data are considered.

25 26 27 **PROCEDURE**

28 29 *Cell maintenance and preparation of the cells for the assay*

30
31 17. Stock cultures of L929 mouse fibroblasts, grown routinely in cell culture flasks, should be
32 maintained and passaged in Dulbecco's Modified Eagle's Medium with 1.0 mM sodium pyruvate
33 (DMEM) containing 10% Fetal Bovine Serum and 2.0 mM L-glutamine (Growth Medium), under
34 normal growth conditions (see paragraph 18). When preparing the cells for the assay, L929 cells at or
35 near confluency are trypsinised, centrifuged and an appropriate cell suspension is prepared and
36 seeded in DMEM containing 1% Fetal Bovine Serum, 50 µg/mL gentamicin and 2.0 mM L-
37 glutamine (Seeding Medium). At the time of the assay itself the cells should be 70 to 80 % confluent
38 in the sensor chamber to allow for accurate pH readings (8). With the CM instrument, this can be
39 achieved by seeding the cells in the capsule cups at a density of $\sim 6 \times 10^5$ cells/cup ($\sim 5.36 \times 10^5$
40 cells/cm²), and incubating them for 16-32 hours under normal growth conditions (see paragraph 18),
41 before use. Prior to the start of the assay, the medium in sensor chambers containing the cultured
42 L929 cells is changed to serum-free, NaHCO₃-free DMEM supplemented to contain 2.0 mM L-
43 glutamine, 50 µg/mL gentamicin, and additional NaCl to preserve osmolarity (NaCl concentration
44 will be increased from 110 mM NaCl to a final concentration of 154 mM NaCl, to substitute for the
45 omission of NaHCO₃) (Low-Buffered Medium).

46 The data provided by the instrument are based on time-dependent changes in pH which occur as a
47 result of cellular metabolism. Use of fully buffered medium would essentially eliminate the ability to
48 detect the necessary level of pH changes.

49

1 18. The L929 cell cultures should be kept in incubators in a humidified atmosphere, at $5 \pm 1\%$
 2 CO_2 and $37 \pm 1^\circ\text{C}$. The cells should be free of contamination by bacteria, viruses, mycoplasma and
 3 fungi.

4
 5 ***Application of the Test and Control Chemicals***

6 19. A fresh stock solution of test chemical should be prepared for each experimental run and
 7 used within 30 minutes of preparation. Test chemicals should be prepared in Low-Buffered Medium.

8
 9 20. A dose range finding assay is performed to establish an appropriate test chemical dose range
 10 for the definitive toxicity test. Solutions at different concentrations are prepared by serial three-fold
 11 dilutions in sterile, Low-Buffered Medium that has been left to equilibrate to room temperature
 12 overnight. The concentrations to be tested in the dose-range finding assay are as follows: 100 mg/mL;
 13 33.3 mg/mL; 11.1 mg/mL; 3.7 mg/mL; 1.23 mg/mL; 0.412 mg/mL and 0.137 mg/mL.. If possible, the
 14 test chemical concentration that results in the reduction of the MRD_{50} value should be calculated from
 15 the dose-range finding assay.

16
 17 21. In the definitive assays seven concentrations are tested. Generally, three concentrations are
 18 chosen below the expected MRD_{50} value, one at approximately the MRD_{50} value, and three above the
 19 expected MRD_{50} . If the test chemical fails to cause 50% toxicity in the dose range finding assay, the
 20 maximum concentration used will generally be 270 mg/mL or less, based on the ability of the test
 21 chemical to form stable single phase solution/suspension in the Low-Buffered Medium during
 22 analysis time (see paragraph 8).

23
 24 22. Once a concentration range which includes the MRD_{50} value has been found, the same range
 25 of concentrations should be tested at least once more, meaning that the final MRD_{50} value is
 26 estimated from the mean of at least two definitive trials A dose range finding trial may be used as a
 27 definitive trial if the requirements for a definitive trial (see paragraph 21) are met.

28
 29 23. The negative control to obtain the basal metabolic rate is Low-Buffered Medium alone. A
 30 concurrent positive control should be used in each experimental run. A solvent control is
 31 recommended when a solvent other than Low-Buffered Medium is used. The maximum solvent
 32 concentration (other than Low-Buffered Medium) should normally be 10%(w/v). The suggested
 33 positive control chemical is Sodium lauryl sulphate (CAS No. 151-21-3) prepared as stock solution in
 34 deionised water (100 mg/mL), and subsequently diluted in Low-Buffered Medium for testing. A dose
 35 range finding assay should be performed once on the positive control to set the appropriate ranges for
 36 the subsequent definitive trials. Historical data for the positive control should be established in each
 37 user laboratory to ensure that the instrument provides similar readings from day-to-day, and to enable
 38 comparing data for different test chemicals tested on different days. These data should not differ
 39 substantially from previously established historical ranges. As a reference, the results compiled by
 40 the Institute for In Vitro Sciences (IIVS) for 629 assays of the positive control (SLS) conducted over
 41 a 12 plus year period as well as for the last 94 of those 629 assays conducted over a period of two
 42 years, are summarised below.

43

Chemical	Dates	No. of Assays	Mean MRD_{50} (mg/mL)	SD	CV (%)
SLS	April, 14 1994 – June 30, 2006	629	0.0799	0.011	14.3

44 Based on the Positive Control data compiled by IIVS this range is 0.0579 – 0.1019 mg/mL.

45 ***Determination of the change in metabolic rate***

46

1 24. Prior to the start of the assay, the medium in sensor chambers containing the cultured L929
2 cells is changed to Low-Buffered Medium (see paragraph 17). The medium flow in the instrument is
3 adjusted and obvious bubbles are cleared. While medium is flowing through the chamber, the pH is
4 stable and governed by the medium. When the flow of medium is stopped, the pH begins to drop in a
5 linear fashion over time.

6
7 25. It should always be ascertained that the instrument and cells are stable before the experiment
8 can begin, since all subsequent data points are interpreted based on the baseline acidification rate
9 (pH/s). Thus, at the beginning of each assay, at least four to five measurements are taken to assess the
10 basal acidification rate (in the case of the validated CM instrument it is measured in $\mu\text{V/s}$), which is
11 used as the negative control for each cell culture. For each sensor chamber, these baseline data points
12 should vary from their mean by no more than 10%, and will be determined just prior to introduction
13 of the first sample dilutions. If the baseline data contain one out of five outlying points that can be
14 explained (e.g., caused by a bubble), it is permissible to delete that data point and use only four for
15 calculations. With the validated CM instrument, baseline rates are expected to fall between 50 and
16 200 $\mu\text{V/s}$ after a stabilization period of approximately 1 hour. The rationale is to verify that the cells
17 in the sensor chamber are sufficient (confluence >70%) to generate a sufficiently high rate to
18 determine the MRD_{50} with a good signal/noise (50 $\mu\text{V/s}$), and to avoid a situation with a confluence
19 >80% together with a high pH drop (200 $\mu\text{V/s}$). If a sensor chamber with cells fails to achieve these
20 ranges it should be discarded and the cells should be replaced.

21
22 26. After the baseline data points have been taken, the cells contained in the chamber undergo
23 cycles of exposure to the test chemical consisting of three phases (exposure, wash-out and
24 measurement). The cycles start from the lowest concentration tested and are repeated for the
25 increasing concentrations of the test chemical in the same cell population. When using the CM
26 instrument, each cycle takes approximately 20 minutes. For standard safety assays the exposure time
27 should be 810 seconds in order to match the experimental conditions for which the main prediction
28 model was established. Longer or shorter exposure times will change the calculated MRD_{50} since
29 toxicity is a function of exposure time. If a different exposure time is used, a conversion algorithm
30 may have to be developed to translate the generated data to validated CM transwell data.

31
32 27. The exposure cycle that should be used with the validated CM instrument, and as specified in
33 the validated CM protocol (INVITTOX Protocol No. 130) (7) is as follows. In the first phase of an
34 exposure cycle, the test chemical is introduced into the sensor for 810 seconds. The nominal medium
35 flow rate is 100 $\mu\text{L/min}$ for the first minute and 20 $\mu\text{L/min}$ for the remaining 12 minutes and 30
36 seconds. During the second phase, which lasts 6 minutes (at a flow rate of 100 $\mu\text{L/min}$), the test
37 chemical is washed out from the sensor chamber using the Low-Buffered Medium not containing the
38 test chemical. In the third phase the flow is stopped (0 $\mu\text{L/min}$) for 25 seconds and the rate of pH
39 change is measured. These cycles (exposure, wash-out and measurement phases) are repeated with
40 increasing concentrations of the test chemical until the highest concentration is reached. If a different
41 exposure cycle is used (e.g. due to the use of a different instrument), a conversion algorithm may
42 have to be developed to translate the generated data to validated CM transwell data.

43 44 *Interpretation of results and Prediction model*

45
46 28. The acidification rates that occur after exposure to each test chemical concentration are
47 calculated and compared to the mean basal acidification rate of the same cells prior to exposure to the
48 test chemical. The percent of control acidification rate is determined by comparing the dose response
49 acidification rate to the basal acidification rate.

50
51 The following equation for the calculation of % control acidification rate should be applied:

$$\% \text{ of control acidification rate} = \frac{\text{acidification rate after exposure to test chemical}}{\text{basal acidification rate}} \times 100$$

The percent of control acidification rates for each concentration are then plotted against the test chemical concentrations. The concentration of the test chemical that results in a 50% reduction in acidification rate is interpolated from the obtained curve and referred to as the MRD₅₀. MRD₅₀ data should be expressed in mg/mL.

29. The cut-off values of MRD₅₀ for predicting chemicals as not classified as irritant, or as ocular corrosives/severe irritants are given below:

Top Down approach: Identification of **severe irritants** (for water soluble chemicals (substances and mixtures))

MRD ₅₀ (mg/mL)	UN GHS C&L[1]	U.S. EPA C&L[3]
> 2 mg/mL	No prediction can be made	No prediction can be made
≤ 2 mg/mL	Category 1	Category I

Bottom up approach: Identification of **non irritants** (for water soluble surfactants and surfactant containing mixtures)

MRD ₅₀ (mg/mL)	UN GHS C&L[1]	U.S. EPA C&L[3]
> 80 mg/mL	N/A	Category IV
≤ 80 mg/mL	N/A	No prediction can be made
> 10 mg/mL	Not Classified	N/A
≤ 10 mg/mL	No prediction can be made	N/A

C&L: classification and labeling; N/A: Not applicable for the particular classification and labelling system

The CM test method is recommended only for the identification of ocular corrosives and severe irritants (UN GHS Category 1) and for the identification of chemicals not classified as irritant (UN GHS Not Classified), within a previously stated applicability domain (see paragraphs 1, 8, 9 and 10).

In this context, the MRD₅₀ cut-off value ≤ 2 mg/mL should be used for the identification of ocular corrosives and severe irritants (for UN GHS) for water soluble chemicals (substances and mixtures) (4, 7, 8), while the MRD₅₀ cut-off value > 10 mg/mL (for UN GHS) or > 80 mg/mL (for U.S. EPA C&L) should be used for the identification of chemicals not classified as irritant for water soluble surfactants and water soluble surfactant-containing mixtures (4, 7, 8).

Acceptance of results

1 30. The test acceptance criteria are as follows:

- 2 *i)* acceptance of test results from a trial are dependent on acceptance of results for the
3 positive control within the trial, which should fall within an acceptable range. A test is
4 considered acceptable if the MRD₅₀ of the positive control falls within 2 standard
5 deviations of the historical mean. The positive control historical mean should be
6 established for each participating laboratory based on at least a minimum number of
7 experiments (each consisting of three definitive trials) statistically defined based on the
8 variability of the positive control. As an example, to establish such historical mean in the
9 VRM, the Institute for In Vitro Sciences (IIVS) compiled results from 629 assays of the
10 positive control (SLS) conducted over a 12 plus year period and based on those (see
11 paragraph 23) this range is 0.0579 – 0.1019 mg/mL.

12 **DATA AND REPORTING**

13 **Data**

14 31. For each run, data from individual replicate measurements (*e.g.* basal acidification rate,
15 acidification rate after exposure to a concentration of the test chemical and calculated % of control
16 acidification rate) should be reported in tabular form. In addition means \pm SD of individual replicate
17 measurements in each run should be reported.
18
19
20

21 **Test Report**

22 32. The test report should include the following information:
23

24 ***Test and Control Chemicals***

- 25 - Chemical name(s) such as the structural name used by the Chemical Abstracts Service
26 (CAS), followed by other names, if known;
- 27 - Chemical CAS number, if known;
- 28 - Purity and composition of the chemical or mixture (in percentage(s) by weight), to the extent
29 this information is available;
- 30 - Physical-chemical properties relevant to the conduct of the study (*e.g.* physical state,
31 volatility, pH, stability, water solubility, chemical class);
- 32 - Treatment of the test/control chemical prior to testing, if applicable (*e.g.* warming, grinding);
- 33 - Storage conditions;

34 ***Justification of the Test Method and Protocol Used***

- 35 - Should include considerations regarding applicability domain and limitations of the test
36 method;

37 ***Test Conditions***

- 38 - Description of cell system used, including certificate of authenticity and the mycoplasma
39 status of the cell line;
- 40 - Details of test procedure used;
- 41 - Test chemical concentration(s) used;
- 42 - Duration of exposure to the test chemical;
- 43 - Description of any modifications of the test procedure;
- 44 - Description of evaluation criteria used;
- 45 - Reference to historical data of the model (*e.g.* negative and positive controls, solvent control,
46 benchmark chemicals, if applicable);

- 1 - Information on the technical proficiency demonstrated by the laboratory;
2
3

4 ***Results***

- 5 - Tabulation of data from individual baseline measurements, test chemicals and positive
6 control for each trial (including individual results, means and SDs);
7 - The derived classification(s) with reference to the prediction model and/or decision criteria
8 used;
9 - Description of other effects observed;

10
11 ***Discussion of the Results***

- 12 - Should include considerations regarding a non-conclusive outcome (paragraph 28: "No
13 prediction can be made") and further testing;

14
15 ***Conclusions***
16
17
18

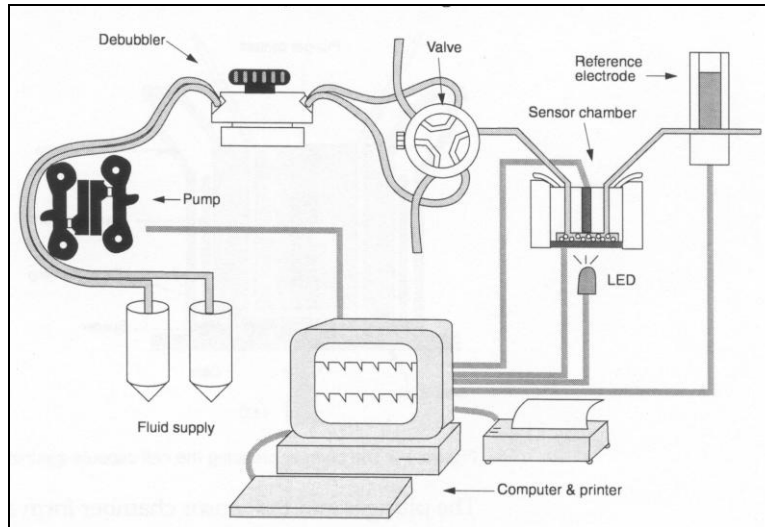
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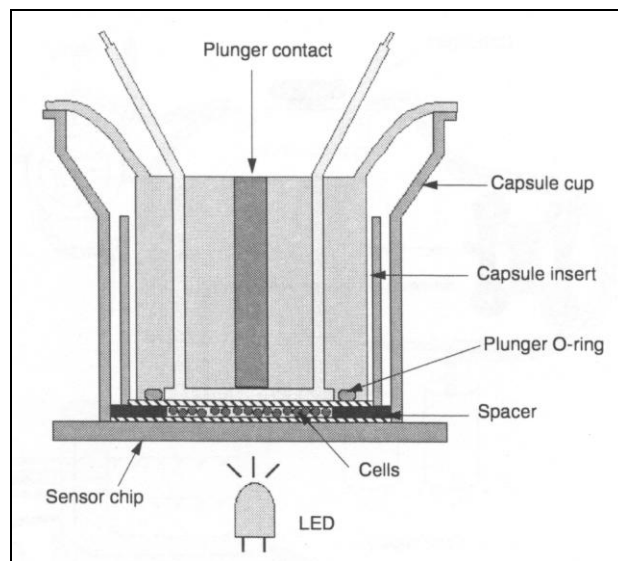
ANNEX I

DIAGRAMS OF THE OPERATING COMPONENTS OF THE CM

The operating components of the CM instrument (diagram taken from the Cytosensor Manual).



The low volume sensor chamber with the transwell in place (diagram taken from the Cytosensor Manual). The capsule insert described in the Cytosensor Manual and which can be seen in this diagram must not be used in the assay (8).



The Cytosensor uses a low volume flow through chamber and a pH-meter to measure the metabolic rate of a cell population. The pH-meter forms the bottom of the flow through chamber and serves as a very sensitive and stable pH meter. The cells are grown on a transwell membrane, which is placed into the sensor chamber and a plunger (with a spacer) pressed down on the membrane to seal it. There is a small medium-filled space between the sensor chip and the bottom of the transwell. The cells are attached to the top of the membrane so

1 that the acid metabolites should pass through the membrane pores to reach the space in the lower part of the
2 chamber. The medium is passed over the cells on the upper side of the membrane.
3

ANNEX II

DEFINITIONS

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5 **Accuracy:** The closeness of agreement between test method results and accepted reference values. It is a
6 measure of test method performance and one aspect of “relevance.” The term is often used interchangeably
7 with “concordance”, to mean the proportion of correct outcomes of a test method.
8

9 **EPA Category I:** Chemicals that produce corrosive (irreversible destruction of ocular tissue) or corneal
10 involvement or irritation persisting for more than 21 days (3).
11

12
13 **False negative rate:** The proportion of all positive chemicals falsely identified by a test method as negative. It
14 is one indicator of test method performance.
15

16 **False positive rate:** The proportion of all negative chemicals that are falsely identified by a test method as
17 positive. It is one indicator of test method performance.
18

19 **GHS (Globally Harmonized System of Classification and Labelling of Chemicals by the United Nation**
20 **(UN)):** A system proposing the classification of chemicals (substances and mixtures) according to
21 standardized types and levels of physical, health and environmental hazards, and addressing corresponding
22 communication elements, such as pictograms, signal words, hazard statements, precautionary statements and
23 safety data sheets, so that to convey information on their adverse effects with a view to protect people
24 (including employers, workers, transporters, consumers and emergency responders) and the environment (1).
25

26 **GHS Category 1:** Production of tissue damage in the eye, or serious physical decay of vision, following
27 application of a test chemical to the anterior surface of the eye, which is not fully reversible within 21 days of
28 application (1).
29

30 **Hazard:** Inherent property of an agent or situation having the potential to cause adverse effects when an
31 organism, system or (sub) population is exposed to that agent.
32

33 **Me-too test:** A colloquial expression for a test method that is structurally and functionally similar to a
34 validated and accepted reference test method. Such a test method would be a candidate for catch-up validation.
35 Interchangeably used with similar test method.
36

37 **Mixture:** Used in the context of the UN GHS (1) as a mixture or solution composed of two or more
38 substances in which they do not react.
39

40 **MRD₅₀:** Metabolic rate decrement of 50%. The concentration of test chemical as weight/volume %, required
41 to reduce the acidification rate by 50%.
42

43 **Negative control:** An untreated replicate containing all components of a test system. This sample is processed
44 before exposure to the test chemical, to assess the basal acidification rate.
45

46 **Not-classified:** Chemicals that are not classified as UN GHS Categories 1, 2A, or 2B or U.S. EPA Categories
47 I, II, or III ocular irritants (1)(3).
48

49 **Ocular corrosive:** (a) A chemical that causes irreversible tissue damage to the eye. (b) Chemicals that are
50 classified as UN GHS Category 1; or U.S. EPA Category I ocular irritants (1)(3).
51

1 **Ocular irritant:** (a) A chemical that produces a reversible change in the eye following application to the
2 anterior surface of the eye; (b) Chemicals that are classified as UN GHS Categories 2A, or 2B; or U.S. EPA
3 Categories II or III ocular irritants (1)(3).
4

5 **Ocular severe irritant:** (a) A chemical that causes tissue damage in the eye following application to the
6 anterior surface of the eye that is not reversible within 21 days of application or causes serious physical decay
7 of vision. (b) Chemicals that are classified as UN GHS Category 1; or U.S. EPA Category I ocular irritants
8 (1)(3).
9

10 **Performance standards (PS):** Standards, based on a validated test method, that provide a basis for evaluating
11 the comparability of a proposed test method that is mechanistically and functionally similar. Included are; (i)
12 essential test method components; (ii) a minimum list of Reference Chemicals selected from among the
13 chemicals used to demonstrate the acceptable performance of the validated test method; and (iii) the
14 comparable levels of accuracy and reliability, based on what was obtained for the validated test method, that
15 the proposed test method should demonstrate when evaluated using the minimum list of Reference Chemicals.
16

17 **Positive control:** A replicate containing all components of a test system and treated with a chemical known to
18 induce a positive response. To ensure that variability in the positive control response across time can be
19 assessed, the magnitude of the positive response should not be excessive.
20

21 **Reference chemicals:** Chemicals selected for use in the validation process, for which responses in the *in vitro*
22 or *in vivo* reference test system or the species of interest are already known. These chemicals should be
23 representative of the classes of chemicals for which the test method is expected to be used, and should
24 represent the full range of responses that may be expected from the chemicals for which it may be used, from
25 strong, to weak, to negative. Different sets of reference chemicals may be required for the different stages of
26 the validation process, and for different test methods and test uses (11).
27

28 **Relevance:** Description of relationship of the test to the effect of interest and whether it is meaningful and
29 useful for a particular purpose. It is the extent to which the test correctly measures or predicts the biological
30 effect of interest. Relevance incorporates consideration of the accuracy (concordance) of a test method (11).
31

32 **Reliability:** Measures of the extent that a test method can be performed reproducibly within and between
33 laboratories over time, when performed using the same protocol. It is assessed by calculating intra- and inter-
34 laboratory reproducibility and intra-laboratory repeatability.
35

36 **Replacement test:** A test which is designed to substitute for a test that is in routine use and accepted for
37 hazard identification and/or risk assessment, and which has been determined to provide equivalent or
38 improved protection of human or animal health or the environment, as applicable, compared to the accepted
39 test, for all possible testing situations and chemicals.
40

41
42 **Solvent control:** An untreated sample containing all components of a test system, including the solvent that is
43 processed with the test chemical-treated and other control samples to establish the baseline response for the
44 samples treated with the test chemical dissolved in the same solvent. When tested with a concurrent negative
45 control, this sample also demonstrates whether the solvent interacts with the test system.
46

47
48 **Substance:** Used in the context of the UN GHS as chemical elements and their compounds in the natural state
49 or obtained by any production process, including any additive necessary to preserve the stability of the product
50 and any impurities deriving from the process used, but excluding any solvent which may be separated without
51 affecting the stability of the substance or changing its composition.

- 1
2 **Surfactant:** Also called surface-active agent, this is a substance, such as a detergent, that can reduce the
3 surface tension of a liquid and thus allow it to foam or penetrate solids; it is also known as a wetting agent.
4
- 5 **Surfactant-based formulations:** In the context of this Test Guideline, it is a formulation containing one or
6 more surfactants at a final concentration of >5%.
7
- 8 **Tiered testing strategy:** A stepwise testing strategy where all existing information on a test chemical is
9 reviewed, in a specified order, using a weight of evidence process at each tier to determine if sufficient
10 information is available for a hazard classification decision, prior to progression to the next tier. If the irritancy
11 potential of a test chemical can be assigned based on the existing information, no additional testing is required.
12 If the irritancy potential of a test chemical cannot be assigned based on the existing information, a step-wise
13 sequential animal testing procedure is performed until an unequivocal classification can be made.
14
- 15 **Validated test method:** A test method for which validation studies have been completed to determine the
16 relevance (including accuracy) and reliability for a specific purpose. It is important to note that a validated test
17 method may not have sufficient performance in terms of accuracy and reliability to be found acceptable for the
18 proposed purpose (11).
19
- 20 **Weight-of-evidence:** The process of considering the strengths and weaknesses of various pieces of
21 information in reaching and supporting a conclusion concerning the hazard potential of a chemical.
22

ANNEX III

PERFORMANCE STANDARDS FOR ASSESSMENT OF PROPOSED SIMILAR OR MODIFIED CYTOSENSOR MICROPHYSIOMETER (CM) TEST METHODS FOR EYE IRRITATION

INTRODUCTION

1. The purpose of Performance Standards (PS) is to provide the basis by which new or modified test methods, both proprietary (*i.e.* copyrighted, trademarked, registered) and non-proprietary can demonstrate to have sufficient reliability and relevance for specific testing purposes. The PS, based on validated and accepted test methods, can be used to evaluate the reliability and relevance of other analogous test methods (colloquially referred to as “me-too” test methods) that are based on similar scientific principles and measure or predict the same biological or toxic effect (9). On the other hand, modified test methods, which propose potential improvements to an approved test method, should be evaluated to determine the effect of the proposed changes on the test method’s performance and the extent to which such changes affect the information available for the other components of the validation process. Depending on the number and nature of the proposed changes, the generated data and supporting documentation for those changes, they should either be subjected to the same validation process as described for a new test method, or, if appropriate, to a limited assessment of reliability and relevance using established PS (11).

2. Similar (me-too) or modified test methods proposed for use under this Test Guideline should be evaluated to determine their reliability and relevance using Reference Chemicals (Table 1). The proposed similar or modified test methods should have reliability, sensitivity, specificity and accuracy values which are comparable or better than those derived from the VRM (CM) and as described in paragraphs 8 to 10 of this Annex (Tables 2 and 3). The reliability of the new or modified test method, as well as its ability to correctly identify non-irritant and irritant chemicals, should be determined prior to its use for testing new chemicals.

3. These PS are based on the ECVAM PS (12) for evaluating the validity of new or modified CM test methods. The PS consists of (11): (i) essential test method components; (ii) recommended reference chemicals (including a set of proposed proficiency chemicals), and; (iii) defined reliability and accuracy values that the proposed test method should meet or exceed.

I. ESSENTIAL TEST METHOD COMPONENTS

4. These consist of essential structural, functional, and procedural elements of a validated test method that should be included in the protocol of a proposed, mechanistically and functionally similar or modified test method. These components include unique characteristics of the test method, critical procedural details, and quality control measures. Adherence to essential test method components will help to assure that a similar or modified proposed test method is based on the same concepts as the corresponding VRM. The essential test method components are described in detail in paragraphs 17 to 30 of the Test Guideline:

- Principle of the test (paragraphs 12 to 16)
- Cell maintenance and preparation of the cells for the assay (paragraphs 17-18)
- Application of the Test and Control Chemicals (paragraphs 19-23)
- Determination of the change in metabolic rate (paragraphs 24-27)
- Interpretation of results and Prediction model (paragraphs 28-29)
- Acceptance of results (paragraph 30)

For specific parameters (e.g. for Table 3 and 4), adequate values should be provided for any new similar or modified test method; these specific values may vary depending on the specific test method..

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II. MINIMUM LIST OF REFERENCE CHEMICALS

5. Reference Chemicals are used to determine if the reliability and relevance of a proposed similar or modified test method, proven to be structurally and functionally sufficiently similar to the VRM, or representing a minor modification of the VRM, are comparable or better than the VRM. The 30 recommended Reference Chemicals listed in Table 1 include substances representing different chemical classes (*i.e.* chemical categories based on functional groups), while a separate table of only the subset of the surfactants has been prepared (Table 2). The substances included in this list comprise 11 UN GHS Category 1, 9 UN GHS Category 2A/2B and 10 "No-Category" chemicals. The substances listed in Table 1 are selected from the substances used in the validation study of the VRM, with regard to chemical functionality and physical state (8). These Reference Chemicals represent the minimum number of chemicals that should be used to evaluate the reliability and relevance of a proposed similar or modified test method able to identify Category 1 and "No-category" chemicals in accordance with the UN GHS (1). The use of these Reference Chemicals for the development/optimization of new similar test methods should be avoided to the extent possible. In situations where a listed substance is unavailable, other substances for which adequate *in vivo* reference data are available could be used, primarily from the substances used in the validation study of the VRM. If desired, additional substances representing other chemical classes and for which adequate *in vivo* reference data are available may be added to the minimum list of Reference Chemicals to further evaluate the accuracy of the proposed test method.

1 **Table 1: Minimum list of Reference Chemicals (to be used in the Top-down approach to identify**
 2 **chemicals as "inducing serious eye damage" (UN GHS Category 1) or as "ocular corrosives or**
 3 **severe irritants" (U.S. EPA Category I),) for determination of reliability, sensitivity, specificity and**
 4 **accuracy values for similar or modified CM-based eye irritation test methods*.**

Substance	CAS Number	Physical state	Conc. Tested (%)	In vivo GHS/EPA	In vitro GHS/EPA ⁴
Benzalkonium chloride ^{1,2}	8001-54-5	Liquid	10	Cat.1/ cat.I	Cat1 / CatI
Sodium lauryl sulphate ^{1,2}	151-21-3	Solid	15	Cat.1/ cat.I	Cat1 / CatI
Dibenzoyl-L-tartaric acid ^{2,3}	2743-38-6	Solid	100	Cat.1/ cat.II	Cat1 / CatI
Cetylpyridinium bromide ^{1,2}	140-72-7	Solid	6	Cat.1/SCNM	Cat1 / CatI
Benzalkonium chloride ^{1,2}	8001-54-5	Liquid	5	Cat.1/ cat.I	Cat1 / CatI
Trichloroacetic acid ²	76-03-9	Solid	30	Cat.1/ cat.I	Cat1 / CatI
Promethazine HCl ^{2,3}	58-33-3	Solid	100	Cat.1/ cat.I	Cat1 / CatI
Sodium perborate, 4H ₂ O ²	10486-00-7	Solid	100	Cat.1/ cat.I	Cat1 / CatI
Triton X-100 ^{1,2}	9002-93-1	Liquid	10	Cat.1/ cat.II	Cat1 / CatI
Sodium hydroxide ²	1310-73-2	Solid	10	Cat.1/ cat.I	NP / NP
Triton X-100 ^{1,2}	9002-93-1	Liquid	5	Cat.2A/ cat.III	NP / NP
Sodium lauryl sulphate ^{1,2}	151-21-3	Solid	3	No Cat./ cat.III	NP / NP
Benzalkonium chloride ^{1,2}	8001-54-5	Liquid	1	Cat.1/ cat.I	NP / NP
Tween 20 ^{1,2}	9005-64-5	Liquid	100	No Cat./ cat.III	NP / NP
Sodium hydroxide ²	1310-73-2	Solid	1	Cat.2B/ cat.III	NP / NP
Ethyl-2-methylacetoacetate ²	609-14-3	Liquid	100	Cat.2B/ cat.III	NP / NP
Methyl ethyl ketone ²	78-93-3	Liquid	100	Cat.2A/ cat.III	NP / NP
Methyl acetate ²	79-20-9	Liquid	100	Cat.2A/ cat.II	NP / NP
Ethanol ^{2,3}	64-17-5	Liquid	100	Cat.2A/ cat.II-III	NP / NP
Acetone ²	67-64-1	Liquid	100	Cat.2A/ cat.II	NP / NP
Isobutanol ²	78-83-1	Liquid	100	Cat.2A/ cat.II	NP / NP
Ammonium nitrate ²	6484-52-2	Solid	100	Cat.2A/ cat.III	NP / NP
Trichloroacetic acid ²	76-03-9	Solid	3	No Cat./ cat.III	NP / NP
Triton X-100 ^{1,2}	9002-93-1	Liquid	1	No Cat./ cat.III	No Cat / NP
Methyl isobutyl ketone ^{2,3}	108-10-1	Liquid	100	No Cat./ cat.III	NP / NP
Ethyl acetate ²	141-78-6	Liquid	100	No Cat./ cat.III	NP / NP
Cetylpyridinium bromide ^{1,2}	140-72-7	Solid	0,1	No Cat./ cat.III	No Cat / CatIV
Glycerol ²	56-81-5	Liquid	100	No Cat./ cat.IV	NP / NP
Propylene glycol ²	57-55-6	Liquid	100	No Cat./ cat.IV	NP / NP
Polyethylene glycol 400 ^{1,2}	25322-68-3	Liquid	100	No Cat./cat.IV	No Cat / CatIV

5 ¹Highlighted substances are surfactants which, together with the rest of the chemicals have been used to estimate the predictive
 6 capacity in a top-down approach. ²In vivo study from ECETOC. ³In vivo study from Gautheron et al., 1994. ⁴Refer to the
 7 prediction model

8 *Number of laboratories where the substances were tested varied from 1-6.

9 NP: means no prediction could be made according to the prediction model as described in the TG, paragraph 29.

10

1 **Table 2: Minimum list of Reference Chemicals (to be used in the Bottom-up approach to**
 2 **identify surfactants or surfactant-based formulations as "not classified for eye irritation"- UN**
 3 **GHS Not Classified or U.S. EPA Category IV) for determination of reliability, sensitivity,**
 4 **specificity and accuracy values for similar or modified CM-based eye irritation test methods*.**

Substance	CAS Number	Physical state	Conc. Tested (%)	In vivo GHS/EPA	In vitro GHS/EPA ⁴
Benzalkonium chloride ^{1,2}	8001-54-5	Liquid	10	Cat.1/ cat.I	NP / NP
Sodium lauryl sulphate ^{1,2}	151-21-3	Solid	15	Cat.1/ cat.I	NP / NP
Cetylpyridinium bromide ^{1,2}	140-72-7	Solid	6	Cat.1/SCNM	NP / NP
Benzalkonium chloride ^{1,2}	8001-54-5	Liquid	5	Cat.1/ cat.I	NP / NP
Triton X-100 ^{1,2}	9002-93-1	Liquid	10	Cat.1/ cat.II	NP / NP
Triton X-100 ^{1,2}	9002-93-1	Liquid	5	Cat.2A/ cat.III	NP / NP
Sodium lauryl sulphate ^{1,2}	151-21-3	Solid	3	No Cat./ cat.III	NP / NP
Benzalkonium chloride ^{1,2}	8001-54-5	Liquid	1	Cat.1/ cat.I	NP / NP
Tween 20 ^{1,2}	9005-64-5	Liquid	100	No Cat./ cat.III	NP / NP
Triton X-100 ^{1,2}	9002-93-1	Liquid	1	No Cat./ cat.III	No Cat / NP
Cetylpyridinium bromide ^{1,2}	140-72-7	Solid	0,1	No Cat./ cat.III	No Cat / CatIV
Polyethylene glycol 400 ^{1,2}	25322-68-3	Liquid	100	No Cat./cat.IV	No Cat / CatIV

5 ¹Highlighted substances are surfactants used to estimate the predictive capacity in a bottom-up approach. ²*In vivo* study
 6 from ECETOC. ³*In vivo* study from Gautheron et al., 1994. ⁴Refer to the prediction model

7 *Number of laboratories where the substances were tested varied from 1-6.

8 NP: means no prediction could be made according to the prediction model as described in the TG, paragraph 29.

9

III. DEFINED RELIABILITY AND ACCURACY VALUES

6. For purposes of establishing the reliability and relevance of proposed similar or modified CM-based test methods to be used by several independent laboratories, all 30 Reference Chemicals listed in Table 1 should be tested in at least three laboratories. It is however essential that all PS-based validation studies are independently assessed by internationally recognized validation bodies, in agreement with international guidelines (11). In each laboratory, three independent classifications should be obtained for each of the reference chemicals for validation purposes. Each classification is obtained from the combination of results of two to three independent and qualified trials, performed with different cell batches and at sufficiently spaced time points. A **qualified trial** consists of a test that meets the test acceptance criteria for the PC, as defined in the corresponding SOP. Otherwise, the trial is considered as non-qualified. The number of qualified trials recommended per tested chemical may be reduced from three to two if statistically/scientifically justified.

7. The calculation of the reliability, sensitivity, specificity and accuracy values of the proposed test method should be done according to the rules described below to ensure that a predefined and consistent approach is used:

1. The within-laboratory reproducibility (WLR) expresses the extent of concordance of the three **classifications** obtained for each Reference Chemical, within each participating laboratory. Each classification is decided using only two/three qualified trials. The number and identity of the Reference Chemicals which have less than three classifications should be reported.
2. The between-laboratory reproducibility (BLR) expresses the extent of concordance among participating laboratories of **final classifications** for each Reference Chemical. The final classification for each Reference Chemical in each participating laboratory is decided based on whether the first two classifications agree with each other. If not, then a third one is obtained which defines the final classification in each laboratory. The number and identity of the Reference Chemicals which have less than three classifications per laboratory should be reported.
3. The calculation of the sensitivity, specificity and accuracy values should be done using all classifications obtained for each Reference Chemical in each laboratory. The calculations should be based on the individual predictions of each classification for each Reference Chemical in each laboratory.

Each laboratory should not produce more than three qualified trials per test chemical for one classification. Excess production of data and subsequent data selection are regarded as not appropriate.

Within-laboratory reproducibility

8. Depending on the regulatory framework and the classification system applied where this method is considered valid, the test method should meet the requirements of either the UN GHS or the U.S. EPA or both classification systems, as described below:

- i) When used in the Bottom-Up Approach (5) to identify surfactants or surfactant-based formulations as "not classified for eye irritation" (UN GHS Not Classified or U.S. EPA Category IV), the concordance of final classifications for the 12 RC (Table 2) obtained in different, independent

1 (minimum of three) experiments within a single participating laboratory must be equal or higher (\geq)
2 than 90%.

3 ii) When used in the Top-Down Approach (5) to identify chemicals as "inducing serious eye
4 damage" (UN GHS Category 1), the concordance of final classifications for the 30 RC (Table 1)
5 obtained in different, independent (minimum of three) experiments within a single participating
6 laboratory must be equal or higher (\geq) than 80%

7
8 *Between-laboratory reproducibility*

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10 9. Depending on the regulatory framework and the classification system applied where this method is
11 considered valid, the test method should meet the requirements of either the UN GHS or the U.S. EPA or both
12 classification systems, as described below:

13
14 i) When used in the Bottom-Up Approach (5) to identify surfactants or surfactant-based formulations
15 as "not classified for eye irritation" (UN GHS Not Classified or U.S. EPA Category IV), the
16 concordance of final classifications for the 12 RC (Table 2) obtained between the different
17 participating laboratories (minimum of three) must be equal or higher (\geq) than 90%.

18 [The actual values for the validated method were: 100% (UN GHS) and 94.44% (U.S. EPA)].

19 ii) When used in the Top-Down Approach (5) to identify chemicals as "inducing serious eye damage"
20 (UN GHS Category 1), the concordance of final classifications for the 30 RC (Table 1) obtained
21 between the different participating laboratories (minimum of three) must be equal or higher (\geq) than
22 80%

23 [The actual value for the validated method was: 87.62% (UN GHS)].

24
25 *Predictive capacity (accuracy)*

26
27 10. The calculation of the accuracy values of the test method should be done on the basis of the RC tested
28 during the validation study. The accuracy values (sensitivity, specificity, false negative rate, false positive rate
29 and overall accuracy) of the proposed similar or modified test method should be done using all qualifying
30 trials obtained for each RC in each laboratory. They should be comparable to those derived from the VRM,
31 taking into consideration additional information relating to relevance in the species of interest. Depending on
32 the regulatory framework and the classification system applied where this method is considered valid, the test
33 method should meet the requirements of either the UN GHS/EU CLP or the U.S. EPA or both classification
34 systems, as described below:

35 i) When used in the Bottom-Up Approach (5) to identify surfactants or surfactant-based formulations as "not
36 classified for eye irritation" (UN GHS Not Classified or U.S. EPA Category IV), the test method should
37 comply with the following capacity predictions (Table 2):

38 The **sensitivity** should be equal to 100% (UN GHS).

39 The **sensitivity** should be equal or higher (\geq) than 94% (U.S. EPA).

40 The **specificity** should be equal or higher (\geq) than 60% (UN GHS).

41 The **specificity** should be equal to 100% (U.S. EPA).

42 **Overall accuracy** should be equal or higher (\geq) than 83.3% (UN GHS)

43 **Overall accuracy** should be equal or higher (\geq) than 94.4% (U.S. EPA).

1
2 **Table 3: Required accuracy values for any similar or modified test method to be considered**
3 **scientifically valid.**

PREDICTIVE CAPACITY – Bottom Up Approach for surfactants					
Sensitivity		Specificity		Overall Accuracy	
GHS	EPA	GHS	EPA	GHS	EPA
100%	94%	60%	100%	83.3%	94.4%
FNR		FPR			
GHS	EPA	GHS	EPA		
0%	6%	40%	0%		

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5
6 ii) When used in the Top-Down Approach (5) to identify chemicals as "inducing serious eye damage" (UN
7 GHS/EU CLP Category 1) or as an "ocular corrosive or severe irritant" (U.S. EPA Category I), the test method
8 should comply with the following capacity predictions (Table 3):

9 The **sensitivity** should be equal or higher (\geq) than 76% (UN GHS)

10 The **sensitivity** should be equal or higher (\geq) than 72.1% (U.S. EPA).

11 The **specificity** should be equal or higher (\geq) than 96.8% (UN GHS)

12 The **specificity** should be equal or higher (\geq) than 89.5% (U.S. EPA).

13 **Overall accuracy** should be equal or higher (\geq) than 89.2% (UN GHS)

14 **Overall accuracy** should be equal or higher (\geq) than 84.7% (U.S. EPA).

15 **Table 4: Required accuracy values for similar or modified CM-based test method to be considered**
16 **valid.**

PREDICTIVE CAPACITY – Top Down Approach for all chemicals					
Sensitivity		Specificity		Overall Accuracy	
GHS	EPA	GHS	EPA	GHS	EPA
76%	72.1%	96.8%	89.5%	89.2%	84.7%
FNR		FPR			
GHS	EPA	GHS	EPA		
24%	27.9%	3.2%	10.5%		

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